### Ouestion 1

What is the most important conclusion derived from the Frye & Edidin experiment?

- a. the plasma membrane thickness is 7 nm.
- b. the plasma membrane is made of 2 layers of phospholipids.
- c. the plasma membrane is fluid at 37 °C.

d the plasma membrane is a sandwich: phospholipids are between 2 layers of proteins.

### Ouestion 2

What is the most important conclusion derived from the Gorter and Grendel experiment?

- a. the plasma membrane thickness is 7 nm.
- b. the plasma membrane is made of 2 layers of phospholipids.
- c. the plasma membrane is fluid at 37 °C.

d the plasma membrane is a sandwich: phospholipids are between 2 layers of proteins.

### Ouestion 3

In an experiment, a blood sample is drawn from a healthy human being.

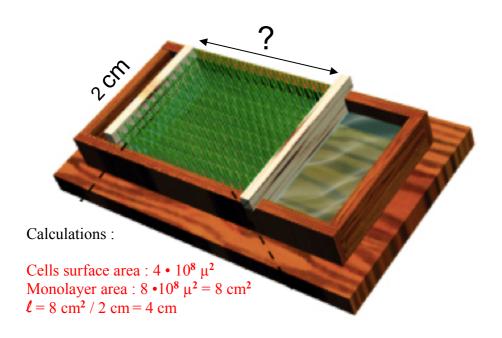
1 μL of this blood is found to contain  $4 \cdot 10^6$  red blood cells (RBC).

Membrane lipids are extracted for the RBC obtained from 1  $\mu$ l of blood and then spread on water in a 2-cm-wide Langmuir through.

What is the distance  $\ell$  between the barriers when the cohesive forces between the lipid molecules signal that a monolayer has been formed?

The surface area of a human red blood cell =  $100 \mu^2$ . Lipid extraction yield (efficiency) = 100 %

l = 4 cm



### Question 4

Phospholipids are purified from human Red Blood Cells and then spread as a monolayer in a Langmuir trough.

Phospholipids from how many Red Blood Cells must be purified to obtain a 1 cm<sup>2</sup> monolayer in the Langmuir trough?

(Assume the purification yield is 100% and a human Red Blood Cell has a 100  $\mu^2$  surface)

 $1 \text{ cm}^2 = 10^8 \,\mu^2$ ; cell surface is half the monolayer area =  $5 \cdot 10^7 \mu^2$ ; cells number =  $5 \cdot 10^5$ 

## Question 5

Let's consider a U-shaped tube with an phospholipids bilayer separating the tube into 2 parts (L and R). This separation is a synthetic membrane made of phospholipids only, no protein, no cholesterol.

You fill both part of the tube (L and R) with 0.1 L of water. The height of the liquid is identical in the L and the R parts.

Table salt (NaCl) and sugar (sucrose) are available for experiments.

Atomic weight of Na +: 23 Atomic weight of Cl- : 35.5

Molecular weight of sucrose: 342

Molecular weight of Sodium Chloride: 23 + 35.3 = 58.5

You dissolve 0.585 g of Sodium Chloride in the 0.1 L of water of the L part.

 $\Diamond$  what is the Na<sup>+</sup>Cl<sup>-</sup> concentration in the L part? \_\_ 100 \_\_\_\_ mM

♦ what is the osmolality in the L part? 200 mOS

100 mOs for Na<sup>+</sup> + 100 mOs for Cl<sup>-</sup>

After 4 hours, you observe that

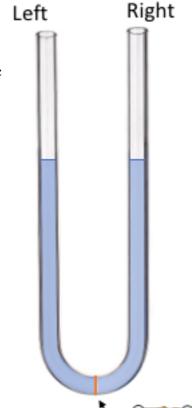
- a. nothing has change, same water levels in the L and R part.
- a. water molecules have moved from the R part to the L part
- a. water molecules have moved from the L part to the R part

You would like to bring back the water levels were they were initially.

To achieve that,

- how much Na<sup>+</sup>Cl<sup>-</sup> should you add to the water in the R part? \_\_\_0.585 \_\_\_\_ g
- how much table sugar (sucrose) should you add to the R part? 6.84 g

Water in part R is isotonic (isoosmolar) with part L when it has 200 mOs. 342 g in 1 L  $\rightarrow$  1 M; 34.2 g in 1 L  $\rightarrow$  100 mM; 3.42 g in 0.1 L  $\rightarrow$  100 mM 6.84 g in 0.1 L  $\rightarrow$  200 mM and 200mOs.



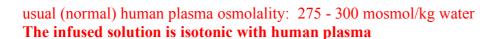
## Question 6

On the site of an accident, an intra-venous (i.v.) line is frequently given to patients. The infused liquid is 0.9 % sodium chloride.

What is the osmolality of 0.9 % sodium chloride ? \_\_\_ 308 mOs\_\_ 58.5 g/L  $\rightarrow$  1M

$$9 \text{ g/L} \rightarrow 0.154 \text{ M} \text{ or } 154 \text{ mM}$$

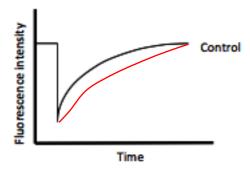
How does this osmolality compare with the osmolality of the human plasma?



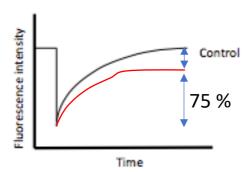


You wish to study how membrane composition affects fluidity using Fluorescence Recovery After Photobleaching (FRAP). On the following graphs, a control FRAP curve has been drawn. Please draw how you predict the **FRAP curve would change for each of the following experiments** 

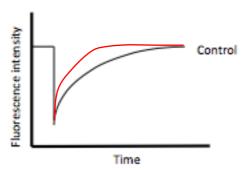
 You add an enzyme which saturates lipid hydrocarbon chains.



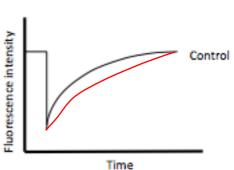
C. You add a drug that prevents 25% of the lipids from diffusing



B. You raise the temperature.

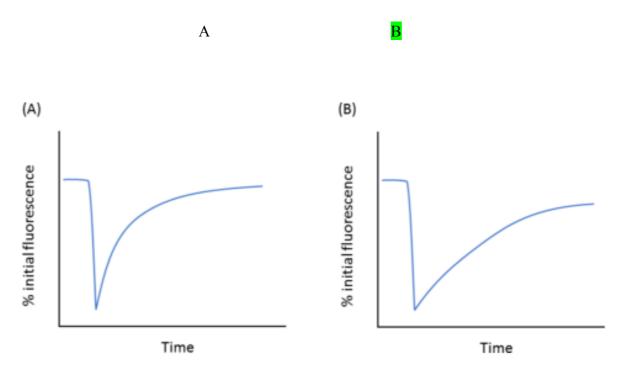


 D. You add an enzyme that makes lipid hydrocarbon chains 10 carbon atoms longer



# Question 8

**Experiment 1.** Two sets of cells are subjected to a FRAP experiment. One cell has a lipid bilayer that includes cholesterol, which makes it less fluid; and the other cell has a membrane made purely of phospholipids. The results are shown below. Which curve, A or B, was obtained from cells with a membrane that includes cholesterol?



**Experiment 2.** Membrane fluidity decreases as the temperature cools. If you conducted two FRAP experiments with the same cell, one experiment at warm temperature, and the other at a cooler temperature, which of the FRAP curves above best represents the cell at the warmer temperature?

A

B

## Notes:

Measures made at 37 °C (> Tm) : cholesterol decreases fluidity (= increases viscosity) Measures made at T < Tm : cholesterol increases fluidity (decreases viscosity)